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## Lactate Dehydrogenase (LDH)

(L-Lactate: NAD<sup>+</sup> oxidoreductase EC 1.1.1.27)

Lactate dehydrogenase (LDH) catalyzes the following reaction:



### ASSAY

The rate of decrease in the absorbancy at 340 nm, resulting from the oxidation of NADH, is a measure of LDH activity.

### REAGENTS

1. 0.1 M Sodium phosphate buffer, pH 7.0.
2. Sodium pyruvate solution, (2.5 mg/ml) in distilled water.
3. NADH solution (5 mg/ml). Dissolve 5 mg NADH, sodium salt in 1.0 ml distilled water. Always prepare fresh.
4. 1% Bovine serum albumin (BSA) solution. Dissolve 1.0 g BSA in 100 ml distilled water. Albumin should be of highest purity.
5. LDH solution (0.5-1.0 U/ml) - Dilute 0.1 ml enzyme suspension to 5 ml with cold 1% BSA solution. Use an aliquot from this stock enzyme solution and dilute to a final concentration of 0.5-1.0 U/ml with cold 1% BSA. This solution must be used as soon as it is prepared and must be made fresh for each run.

### PROCEDURE

1. Set spectrophotometer (equipped with strip chart recorder and temperature control) at 340 nm and 25°C.
2. Into the cuvette, pipette the following:  
0.1 M Phosphate buffer, pH 7.0 2.7 ml  
Sodium pyruvate 0.1 ml  
NADH 0.05 ml
3. Mix and incubate at 25°C for 5 minutes.
4. Transfer the cuvette to the spectrophotometer and record the blank rate for 2-3 minutes.
5. Pipette 0.1 ml fresh enzyme solution (0.5-1.0 U/ml). Mix and monitor the reaction for 5-10 minutes at 340 nm.
6. Calculate  $\Delta E_{340 \text{ nm/min}}$



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#### CALCULATION

$$\text{Activity (U/mg)} = \frac{(\Delta E_{304nm/min}) (\text{Total Vol.}) (\text{Enz. Diln.})}{(6.22) (\text{Enz. Vol.}) (\text{mg Enz. /ml})}$$

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